

COMPOSITION OF THE OIL FROM WASTE TIRES.

1. Fraction boiling at up to 160 °C

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Abstract. The qualitative and quantitative composition of the oil fraction of waste tires boiling at up to 160 °C was determined using capillary GC with OV-101 and SW 10 columns and GC/MS. Over a hundred components (paraffins, naphthenes, olefins, monoterpenes, aromatic hydrocarbons) representing more than 96% of the fraction were identified by means of retention indices and mass spectra. The main components in this oil fraction were aromatic hydrocarbons (27%) and dipentene (12%).

Key words: waste tires, oil, fraction boiling at up to 160 °C, composition, GC, GC/MS.

As part of our research into the utilization problems of waste tires we analysed the qualitative and quantitative composition of oil in radial tires. This paper presents the results obtained for the first fraction, which boils at up to 160 °C.

EXPERIMENTAL

Material

The oil from waste tires was obtained by thermal destruction at 520 °C of tire scraps in a laboratory retort [1]. The total yield of oil from waste tires was 41.2%. The waste tire oil was rectified on the APH-2 apparatus. The fraction boiling at up to 160 °C constituted 19.7% from the total rubber oil [1].

Analysis

The analysis of waste tire oil was performed on a Chrom-5 gas chromatograph (Laboratorni Pristroje, Prague, Czech Republic) equipped with a

flame ionization detector. Helium was used as a carrier gas with a splitting ratio of about 1:150. A Hewlett-Packard Model 3390A integrator was applied for data processing. Table 1 specifies the fused silica columns with bonded stationary phases and the conditions of analyses.

Table 1. Capillary columns used and operating conditions

| Parameter | Stationary phase | |
|--|-------------------------------|----------------------------------|
| | Polymethyl siloxane OV-101 | Polyethylene glycol 20M SW 10 |
| Column length, m | 50 | 60 |
| Column i.d., mm | 0.20 | 0.32 |
| Stationary phase film thickness, μm | 0.50 | 0.25 |
| Number of plates for <i>n</i> -decane at 90°C | 145 000 | 300 000 |
| Injector temperature, °C | 160 | 160 |
| Helium flow rate, ml/min | 0.25–0.28 | 0.87–1.00 |
| Column temperature, °C | 6 min at 30, then 30–100 | 10 min at 40, then 40–130 |
| Programming rate, °C/min | 1 | 2 |

The mass-spectrometric analyses were carried out on a Hitachi-M 80 B gas chromatograph double focussing mass-spectrometer (ionization voltage 70 eV). SPB-1 capillary column (30 m \times 0.25 mm i.d.) and the temperature program 60°C for 5 min and then programmed from 60 to 150°C at 5°/min were used.

The individual compounds in the oil fraction were identified by comparing their retention indices (RI) determined on the temperature programming condition with authentic data that were either determined in our laboratory or obtained from the literature [2–13]. The results were confirmed by GC/MS. The reproducibility of RI expressed in terms of the standard deviation was below 2 index units.

The quantitative composition of components in the oil fraction was calculated using their peak areas without any correction for the relative response factor. The results presented are mean values of three injections.

RESULTS AND DISCUSSION

The complex nature of the oil fraction of waste tires boiling at up to 160°C is demonstrated by the large number of peaks in the chromatogram (Fig. 1). The identified components (103 peaks) and their concentrations in the fraction are listed in Table 2.

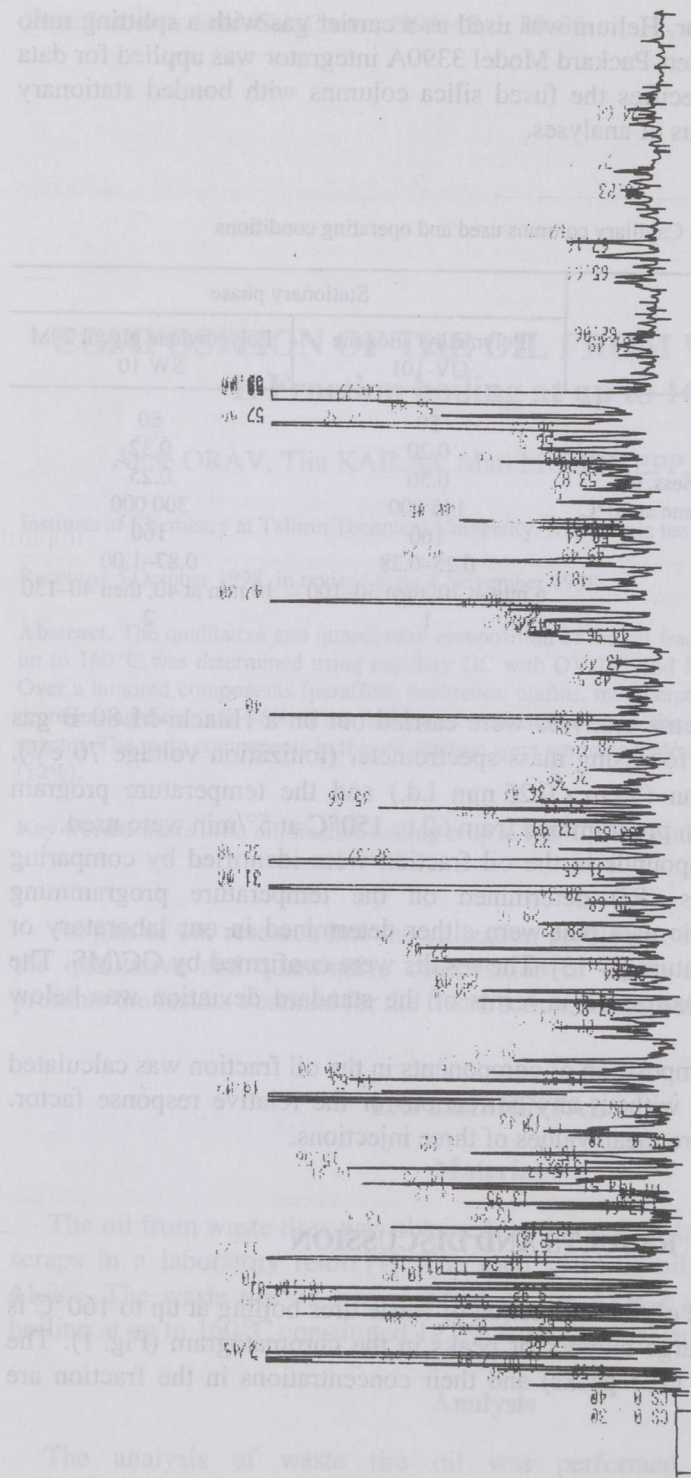


Fig. 1. Gas chromatogram of waste tire oil fraction boiling at up to 160 °C on OV-101 capillary column. For peak identification see Table 2.

Table 2. Composition of the waste tire oil fraction boiling at up to 160 °C and identification data

| Peak No. | Component | RI on OV-101 | Concentration, % | Identification |
|----------|--|--------------|------------------|----------------|
| 1. | 2-Methylpropene | | 0.41 | MS |
| 2. | 2,2-Dimethylpropane | <500 | 0.19 | GC |
| 3. | 3-Methylbutane | <500 | 0.27 | GC |
| 4. | 2-Methylbutane | <500 | 0.47 | GC |
| 5. | 2-Methyl-1-butene | <500 | 1.53 | GC |
| | Acetone | | | GC, MS |
| 6. | <i>n</i> -Pentane | 500 | 5.37 | GC |
| | Isoprene | | | GC, MS |
| 7. | 2-Methyl-2-butene | 522 | 4.16 | GC |
| | 2-Pentene | | | MS |
| 8. | 2,2-Dimethylbutane | 536 | 0.18 | GC |
| 9. | 4-Methyl-1-pentene | 554 | 0.36 | GC, MS |
| 10. | 3-Methyl-1-pentene | 559 | 0.39 | GC, MS |
| 11. | 2,3-Dimethyl-1-butene | 564 | 0.17 | GC |
| | Cyclopentane | | | GC |
| | 2,3-Dimethylbutane | | | GC |
| 12. | 2-Methylpentane | 571 | 0.67 | GC |
| | 4-Methyl- <i>trans</i> -2-pentene | | | GC, MS |
| 13. | 3-Methylpentane | 583 | 0.21 | GC |
| 14. | 1-Hexene | 589 | 0.89 | GC, MS |
| | 2-Methyl-1-pentene | | | GC |
| 15. | Not identified | 594 | 0.35 | |
| 16. | <i>n</i> -Hexane | 600 | 0.61 | GC |
| 17. | 2-Methyl-2-pentene | 607 | 0.91 | GC |
| | 2,3-Dimethyl-1,3-butadiene | | | GC |
| 18. | <i>cis</i> -2-Hexene | 610 | 0.74 | GC |
| | 3-Methyl- <i>cis</i> -2-pentene | | | GC, MS |
| | 2,3-Dimethyl-2,3-butadiene | | | MS |
| 19. | <i>trans</i> -3-Methyl-2-pentene | 620 | 1.04 | GC |
| | <i>trans</i> -4,4-Dimethyl-2-pentene | | | GC |
| 20. | Methylcyclopentane | 626 | 0.47 | GC |
| | 2,2-Dimethylpentane | | | GC, MS |
| 21. | 2,2,3-Trimethylbutane | 634 | 0.58 | GC |
| | 1,3-Cyclohexadiene | | | MS |
| 22. | 2,3,3-Trimethyl-1-butene | 637 | 0.48 | GC |
| 23. | <i>cis</i> -4,4-Dimethyl-2-pentene | 643 | 0.24 | GC |
| | 2,4-Dimethyl-1-pentene | | | GC |
| 24. | Benzene | 649 | 3.02 | GC, MS |
| | 2,4-Dimethyl-2-pentene | | | GC |
| 25. | 2,3-Dimethyl-1-pentene | 658 | 0.19 | GC |
| | Cyclohexane | | | GC |
| 26. | <i>trans</i> -4-Methyl-2-hexene | 664 | 0.30 | GC |
| 27. | 1,1-Dimethylcyclopentane | 673 | 0.52 | GC |
| 28. | <i>cis</i> -3,4-Dimethyl-2-pentene | 677 | 0.32 | GC |
| | Cyclohexene | | | GC, MS |
| | 3-Methylhexane | | | GC, MS |
| 29. | 2-Methyl-1-hexene | 686 | 0.35 | GC |
| | <i>trans</i> -1,3-Dimethylcyclopentane | | | GC |

Table 2 continued

| Peak No. | Component | RI on OV-101 | Concentration, % | Identification |
|----------|--|--------------|------------------|----------------|
| 30. | 1-Heptene | 689 | 0.53 | GC, MS |
| | 2,2,4-Trimethylpentane | | | GC |
| | <i>cis</i> -3-Methyl-3-hexene | | | GC |
| 31. | 2-Methyl-2-hexene | 697 | 0.70 | GC, MS |
| | <i>cis</i> -3-Heptene | | | GC |
| 32. | <i>n</i> -Heptane | 700 | 1.17 | GC, MS |
| | <i>cis</i> -3-Methyl-2-hexene | | | GC |
| 33. | <i>trans</i> -2-Heptene | 708 | 1.03 | GC |
| | 2,3-Dimethyl-2-pentene | | | GC |
| | 2,5-Norbornadiene | | | GC |
| | Diisobutylene | | | MS |
| 34. | <i>cis</i> -2-Heptene | 713 | 0.21 | GC |
| | 2,4,4-Trimethyl-1-pentene | | | GC, MS |
| 35. | Methylcyclohexane | 717 | 0.81 | GC |
| | <i>cis</i> -1,2-Dimethylcyclopentane | | | GC |
| 36. | Ethylcyclopentane | 727 | 0.37 | GC |
| | 2,4,4-Trimethyl-2-pentene | | | GC, MS |
| | 2-Norbornene | | | GC |
| 37. | 2,5-Dimethylhexane | 732 | 0.53 | GC |
| | Dimethylsulphone | | | MS |
| 38. | 2,4-Dimethylhexane | 732 | 0.48 | GC |
| | 2,2,3-Trimethylpentane | | | GC |
| 39. | 4-Methylcyclohexene | 734 | 0.12 | GC, MS |
| | 2,5-Heptadiene | | | MS |
| 40. | 2,3-Dimethyl-1-hexene | 751 | 3.09 | GC |
| | <i>trans</i> -2-Methyl-3-heptene | | | GC |
| | 2-Methyl-2,4-hexadiene | | | MS |
| 41. | Toluene | 754 | 6.80 | GC, MS |
| 42. | 2,3,3-Trimethylpentane | 756 | 0.97 | GC, MS |
| 43. | 2-Methyl,3-ethylpentane | 759 | 0.20 | GC |
| | 2,5-Dimethyl-2-hexene | | | GC |
| | 1,1,2-Trimethylcyclopentane | | | GC |
| | 2-Methyl,3-ethylpentane | | | GC |
| | 2,3-Dimethylhexane | | | GC |
| 44. | <i>trans</i> -4-Methyl-2-heptene | 763 | 0.94 | GC |
| | 3,5,5-Trimethyl-1-hexene | | | GC |
| | 1-Methylcyclohexene | | | GC, MS |
| 45. | 3,4-Dimethylhexane | 767 | 0.49 | GC |
| | 3,5,5-Trimethyl-1-hexene | | | GC |
| 46. | 3-Methyl-3-ethylpentane | 770 | 0.17 | GC |
| | <i>cis</i> -1,2, <i>trans</i> -2,4-Tetramethylcyclopentane | | | GC |
| | 2,2,4,4-Tetramethylcyclopentane | | | GC |
| 47. | 2,3,4-Trimethyl-2-pentene | 773 | 0.81 | GC |
| | 1, <i>cis</i> -3-Dimethylcyclopentane | | | GC |
| | 3-Methylheptane | | | GC |
| | 3-Ethylhexane | | | GC |
| 48. | <i>trans</i> -1,4-Dimethylcyclohexane | 784 | 0.18 | GC |
| | 2,2,5-Trimethylhexane | | | GC |
| | 2-Methyl-1-heptene | | | GC |

Table 2 continued

| Peak No. | Component | RI on OV-101 | Concentration, % | Identification |
|----------|---------------------------------|--------------|------------------|----------------|
| 49. | 2-Methyl-2-heptene | 786 | 0.20 | GC |
| 50. | 1-Octene | 788 | 0.35 | GC, MS |
| 51. | 2,3-Dimethyl-2-hexene | 796 | 0.38 | GC |
| | <i>cis</i> -4-Octene | | | GC, MS |
| | <i>trans</i> -3-Octene | | | GC, MS |
| 52. | <i>n</i> -Octane | 800 | 0.58 | GC, MS |
| 53. | <i>trans</i> -2-Octene | 804 | 0.85 | GC |
| | 2,6-Dimethyl-2,4-hexadiene | | | MS |
| 54. | <i>cis</i> -2-Octene | 811 | 0.15 | GC |
| | 2,4-Octadiene | | | MS |
| | Alkyl imidazol | | tr. | MS |
| | Ethyl phenol | | tr. | MS |
| | Dimethyl phenol | | tr. | MS |
| 55. | 2,2-Dimethylheptane | 818 | 1.16 | GC |
| | 2,2,3,4-Tetramethylpentane | | | GC |
| 56. | 2,4-Dimethylheptane | 823 | 0.85 | GC |
| 57. | 4,4-Dimethylheptane | 826 | 0.54 | GC |
| | 2,6-Dimethylheptane | | | GC, MS |
| 58. | Ethylcyclohexane | 834 | 0.44 | GC |
| | 3,3-Dimethylheptane | | | GC |
| 59. | 2,3-Dimethyl-1-heptene | 842 | 0.30 | GC |
| 60. | Ethylbenzene | 847 | 4.24 | GC |
| 61. | <i>cis</i> -4-Methyl-2-octene | 850 | 0.16 | GC, MS |
| 62. | 1,3-Dimethylbenzene | 857 | 5.46 | GC, MS |
| | 1,4-Dimethylbenzene | | | GC, MS |
| | <i>trans</i> -4-Methyl-2-octene | | | GC |
| 63. | 2-Methyloctane | 862 | 0.40 | GC |
| 64. | 3-Methyloctane | 867 | 0.46 | GC, MS |
| 65. | 2-Methyl-2-octene | 873 | 0.56 | GC, MS |
| 66. | Styrene | 875 | 0.19 | GC, MS |
| 67. | 1,2-Dimethylbenzene | 877 | 0.97 | GC, MS |
| | 2,3-Dimethyl-2-heptene | | | GC |
| 68. | 2-Methyl-1-octene | 880 | 0.64 | GC |
| 69. | Not identified | 883 | 0.36 | |
| 70. | 3,5-Dimethylbenzenemethanol | 885 | 0.32 | MS |
| 71. | 1-Nonene | 890 | 0.25 | GC |
| 72. | <i>trans</i> -4-Nonene | 894 | 0.19 | GC |
| | <i>cis</i> -3-Nonene | | | GC |
| 73. | <i>n</i> -Nonane | 900 | 0.55 | GC |
| 74. | <i>trans</i> -2-Nonene | 903 | 0.35 | GC |
| 75. | Isopropylbenzene | 908 | 1.74 | GC |
| | 2,2-Dimethyloctane | | | GC |
| | <i>cis</i> -2-Nonene | | | GC |
| 76. | 2,2-Dimethyloctane | 918 | 0.18 | GC |
| 77. | 3,5-Dimethyloctane | 924 | 0.38 | GC |
| 78. | 1,2-Dimethyl-1-octene | 929 | 0.37 | GC |
| 79. | <i>n</i> -Propylbenzene | 939 | 0.36 | GC, MS |
| | 3,3-Dimethyloctane | | | GC |

Table 2 continued

| Peak No. | Component | RI on OV-101 | Concentration, % | Identification |
|----------|---------------------------------|--------------|------------------|----------------|
| 80. | Propylcyclohexane | 944 | 0.71 | GC, MS |
| | <i>cis</i> -4-Methyl-2-nonene | | | GC |
| 81. | 1-Methyl,3-ethylbenzene | 947 | 0.48 | GC, MS |
| 82. | 1-Methyl,4-ethylbenzene | 949 | 2.78 | GC |
| 83. | 1,3,5-Trimethylbenzene | 954 | 0.28 | GC, MS |
| | <i>trans</i> -4-Methyl-2-nonene | | | GC |
| 84. | 2,3-Dimethyl-2-octene | 961 | 0.09 | GC |
| 85. | 2-Methylnonane | 963 | 0.25 | GC |
| 86. | 1-Methyl,2-ethylbenzene | 964 | 1.41 | GC |
| | 2-Methyl-2-nonene | | | GC |
| | α -Methylstyrene | | | GC |
| 87. | 3-Methylnonane | 970 | 0.25 | GC |
| 88. | Not identified | 973 | 0.24 | |
| 89. | <i>tert</i> -Butylbenzene | 975 | 0.33 | GC |
| | 2-Butyl-1-hexene | | | GC |
| 90. | 1,2,4-Trimethylbenzene | 976 | 0.94 | GC, MS |
| | Alkyltiophene | | | MS |
| | 2-Methylstyrene | | | GC |
| | 3-Methylstyrene | | | GC |
| | 2-Methyl-1-nonene | | | GC |
| 91. | 4-Methylstyrene | 982 | 0.43 | GC |
| 92. | 2-Ethyl-1-octene | 987 | 0.22 | MS |
| 93. | 1-Decene | 988 | 0.25 | GC, MS |
| 94. | <i>cis</i> -5-Decene | 990 | 0.19 | GC |
| 95. | <i>trans</i> -4-Decene | 992 | 0.27 | GC |
| | <i>cis</i> -4-Decene | | | GC |
| | <i>trans</i> -5-Decene | | | GC |
| 96. | <i>cis</i> -3-Decene | 993 | 0.26 | GC |
| 97. | <i>trans</i> -3-Decene | 995 | 0.51 | GC |
| 98. | <i>n</i> -Decane | 1000 | 0.30 | GC, MS |
| | 1,2,3-Trimethylbenzene | | | GC, MS |
| 99. | <i>trans</i> -2-Decene | 1003 | 0.46 | GC |
| | 1-Methyl,3-isopropylbenzene | | | GC |
| 100. | 1-Methyl,4-isopropylbenzene | 1005 | 3.42 | GC, MS |
| 101. | Terpinene | 1008 | 0.64 | GC, MS |
| | <i>cis</i> -2-Decene | | | GC |
| 102. | Dipentene* | 1018 | 12.60 | GC, MS |
| 103. | 1-Methyl,2-isopropylbenzene | 1024 | 0.22 | GC |
| | Indene | | | GC, MS |
| | Alkyltiophene | | | MS |
| 104. | 1,3-Dimethyl,4-ethylbenzene | 1063 | 0.23 | GC |
| | 2-Methyldecane | | | GC |
| 105. | 1,2-Dimethyl,4-ethylbenzene | 1070 | 0.46 | GC |
| | 3-Methyldecane | | | GC |
| 106. | 1,3-Dimethyl,2-ethylbenzene | 1075 | 0.41 | GC |
| | Terpinolene | | | GC, MS |
| | Total | | 97.55 | |

* 1-Methyl-4-isopropenyl-1-cyclohexene.

The GC and GC/MS identification showed that the light fraction of tire waste oil contained various groups of hydrocarbons (paraffins, olefins, naphthenes, terpenic and aromatic hydrocarbons) and some oxygen and sulphuric compounds. The main group of components in this fraction consisted of aromatic hydrocarbons (Table 3). The total amount of unsaturated hydrocarbons (monoterpenes, alkadienes, isoalkenes, *n*-alkenes, cycloalkenes, and cycloalkadienes) was very high (45%), isoalkanes made up 9%, while *n*-alkanes and cyclanes occurred only in small quantities (<3%) in the light fraction of tire oil.

Table 3. Group content of the rubber waste oil fraction boiling at up to 160 °C

| Group of components | Concentration, % | |
|----------------------------------|------------------|-------|
| | OV-101 | SW 10 |
| <i>n</i> -Alkanes | 2.9 | |
| Isoalkanes | 9.0 | |
| <i>n</i> -Alkenes | 8.6 | |
| Isoalkenes | 11.7 | |
| Alkadienes | 9.6 | |
| Cyclanes | 2.7 | |
| Cycloalkenes and cycloalkadienes | 1.5 | |
| Monoterpenes | 13.6 | 12.5 |
| Arenes | 34.3 | 26.6 |
| Oxygen compounds | 1.9 | |
| Sulphuric compounds | 0.8 | |
| Total | 96.6 | |

To control the concentration of aromatic hydrocarbons in the oil fraction a SW 10 capillary column was used. On the polar SW 10 stationary phase the retention times of arenes, which are the most polar compounds, are much higher than those of the other hydrocarbon groups compared with nonpolar OV-101 phase, where the peaks of arenes and other hydrocarbon groups can overlap. For that reason the amount of arenes in the fraction boiling at up to 160 °C calculated on the SW 10 column was somewhat smaller (27%) than the sum on the OV-101 column (34%).

The major constituent in the fraction studied was dipentene, which formed 12% of the oil fraction. Isoprene, 2-pentene, benzene, toluene, 2-methyl-2,4-hexadiene, ethylbenzene, dimethylbenzenes, and *p*-cumene (1-methyl-4-isopropylbenzene) were found in quantities over 3%.

CONCLUSIONS

The oil fraction boiling at up to 160°C constitutes about 20% of the total waste tire oil. It is a very complicated mixture of hydrocarbons. The main groups of components of this fraction are aromatic and unsaturated hydrocarbons. The major individual component in this mixture is dipentene.

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SÕIDUAUTO KASUTATUD RADIAALKUMMI ÕLI KOOSTIS

1. Kuni 160°C juures keev fraktsioon

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Kapillaargaasikromatograafia ja massispektromeetria meetoditega on määratud sõiduauto kasutatud radiaalkummi termilisel lagundamisel saadud õli kuni 160°C juures keeva fraktsiooni kvalitatiivne ja kvantitatiivne koostis. Kummiõli kerges fraktsioonis identifitseeriti üle saja komponendi, mis kokku moodustasid 96% õli koostisest. Kõige rohkem oli aromaatseid ja küllastumata süsivesinikke. Tsüklilisi süsivesinikke (sealhulgas monoterpene) sisaldas õli 18%, *n*-alkaane oli alla 3%. Enam kui 5% leidis uuritud fraktsioonis dipenteeni, isopreeni ja tolueni.